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# Pyridine-4-carbaldehyde azine 

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In the crystal structure of the title compound, $\mathrm{C}_{12} \mathrm{H}_{10} \mathrm{~N}_{4}$, the pyridine ring makes a dihedral angle of $1.12(9)^{\circ}$ with the mean plane of the complete almost planar and crystallographically centrosymmetric molecule. There are stacks of parallel molecules along the $a$-axis direction, with alternate stacks having a herring-bone arrangement relative to each other and an interplanar spacing of $3.551 \AA$.

## Comment

The title compound, (I), is known not only as a coordinating reagent to determine the concentration of the $\mathrm{Fe}^{\mathrm{II}}$ ion in solution (Luque de Castro \& Valcarcel, 1978), but also as a precursor in the atactic polymerization of 4-vinylpyridine (Biedermann et al., 1972). In addition, this Schiff base ligand derived from hydrazine finds a wide range of applications in coordination chemistry owing to its polydentate chelating ability (Tarafder \& Khan, 1991). Although the synthesis and characterization of 4,4-azinodimethyldipyridine and the crystal structure of its diperchlorate salt, $\mathrm{C}_{12} \mathrm{H}_{10} \mathrm{~N}_{4}{ }^{2+} \cdot 2 \mathrm{ClO}_{4}{ }^{-}$, have been reported (Chen et al., 1997), the crystal structure of 4,4'-azinodimethyldipyridine or pyridine-4-carbaldehyde azine, (I), has not been reported. The title compound acts as a potential bridging ligand in the study of supramolecular chemistry and can coordinate to metal ions to form multidimensional structures.

(I)

There is half a molecule in the asymmetric unit and the other half is inversion related. The bond distances and angles agree with those of the perchlorate salt (Chen et al., 1997). The $\mathrm{C} 6=\mathrm{N} 2$ distance is shorter than 1.285 (7) $\AA$ and the bond
angles $\mathrm{C} 6-\mathrm{N} 2-\mathrm{N} 2^{\mathrm{i}}$ [symmetry code: (i) $1-x, 2-y, 1-z$ ] and $\mathrm{C} 1-\mathrm{C} 6-\mathrm{N} 2$ are slightly smaller than those of 112.2 (2) and $120.7(2)^{\circ}$, respectively, reported for $2,2^{\prime}$-azinodimethyldiphenol (Xu et al., 1994). The molecule adopts an antiperiplanar conformation with respect to the $\mathrm{N} 2-\mathrm{N} 2^{\mathrm{i}}$ bond. Atoms C6 and C6 ${ }^{i}$ are in a trans orientation with respect to the N2$\mathrm{N} 2^{\mathrm{i}}$ bond. The pyridine ring makes a dihedral angle of $1.12(9)^{\circ}$ with the mean plane of the complete almost planar molecule. There are stacks of parallel molecules along the $a$ axis direction, with alternate stacks having a herring-bone arrangement relative to each other with an interplanar spacing of $3.551 \AA$.

## Experimental

The title compound was prepared by stirring hydrazine hydrate ( $80 \%$ ) and pyridine-4-carboxaldehyde under flowing $\mathrm{N}_{2}$ in anhydrous EtOH solution at 353 K for 6 h . The solution was then cooled and filtered. The filtrate was evaporated at room temperature for a few weeks and orange block-shaped crystals were obtained.

## Crystal data

$\mathrm{C}_{12} \mathrm{H}_{10} \mathrm{~N}_{4}$
$M_{r}=210.24$
Monoclinic, $P 2_{{ }_{1}} / c$
$a=3.8515$ (3) $\AA$ 。
$b=11.0201$ (8) $\AA$
$c=12.7303$ (9) $\AA$
$\beta=92.312$ (2) ${ }^{\circ}$ 。
$V=539.88(7) \AA^{3}$
$Z=2$

## Data collection

Siemens SMART CCD area-
detector diffractometer

## $\omega$ scans

3584 measured reflections
1325 independent reflections 1065 reflections with $I>2 \sigma(I)$

## Refinement

Refinement on $F^{2}$

$$
\begin{aligned}
& w=1 /\left[\sigma^{2}\left(F_{o}{ }^{2}\right)+(0.0958 P)^{2}\right. \\
& +0.1438 P] \\
& \text { where } P=\left(F_{o}{ }^{2}+2 F_{c}^{2}\right) / 3 \\
& (\Delta / \sigma)_{\max }<0.001 \\
& \Delta \rho_{\max }=0.31 \mathrm{e}_{\mathrm{m}}{ }_{\circ}^{-3} \\
& \begin{array}{l}
\Delta \rho_{\text {max }}=0.31 \mathrm{e}^{-3} \AA_{\text {min }}=-0.20 \mathrm{e}^{-3}
\end{array}
\end{aligned}
$$

## Table 1

Selected geometric parameters ( $\AA{ }^{\circ}{ }^{\circ}$ ).

| $\mathrm{N} 1-\mathrm{C} 4$ | $1.332(3)$ | $\mathrm{N} 2-\mathrm{C} 6$ | $1.266(3)$ |
| :--- | :--- | :--- | :--- |
| $\mathrm{N} 1-\mathrm{C} 3$ | $1.337(3)$ | $\mathrm{N} 2-\mathrm{N} 2^{\mathrm{i}}$ | $1.407(3)$ |
|  |  |  |  |
| $\mathrm{C} 4-\mathrm{N} 1-\mathrm{C} 3$ | $116.7(2)$ | $\mathrm{C} 6-\mathrm{N} 2-\mathrm{N} 2^{\mathrm{i}}$ | $112.2(2)$ |

Symmetry code: (i) $1-x, 2-y, 1-z$.
All the H atoms were located from the difference Fourier map and were refined isotropically.

Data collection: SMART (Siemens, 1996); cell refinement: SAINT (Siemens, 1996); data reduction: SAINT; program(s) used to solve structure: SHELXTL (Sheldrick, 1997); program(s) used to refine
structure: SHELXTL; software used to prepare material for publication: SHELXTL and PLATON (Spek, 1990).

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## References

Biedermann, H. G., Obwandner, J. \& Wichmann, K. (1972). Z. Naturforsch. Teil B, 27, 1332-1335.
Chen, W., Liu, C. M., Li, D. G. \& You, X. Z. (1997). Acta Cryst. C53, 14991501.

Luque de Castro, M. D. \& Valcarcel, M. (1978). Anal. Lett. A11, 1-12.
Sheldrick, G. M. (1997). SHELXTL Software Reference Manual. Version 5.1. Bruker AXS Inc., Madison, Wisconsin, USA.
Siemens (1996). SMART and SAINT. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.
Spek, A. L. (1990). Acta Cryst. A46, C-34.
Tarafder, M. T. H. \& Khan, A. R. (1991). Polyhedron, 10, 819-822.
Xu, X.-X., You, X.-Z. \& Sun, Z.-F. (1994). Acta Cryst. C50, 1169-1171.

